

# Produktinformation



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# PRODUCT INFORMATION



## Aspirin-d<sub>4</sub> Item No. 18243

CAS Registry No.: 97781-16-3

Formal Name: 6-(acetyloxy)-benzoic-2,3,4,5-d<sub>4</sub> acid

Synonym: Acetylsalicylic Acid-d<sub>4</sub>

MF:  $C_9H_4D_4O_4$ FW: 184.2 ≥95% Aspirin **Chemical Purity:** 

Deuterium

 $\geq$ 99% deuterated forms (d<sub>1</sub>-d<sub>4</sub>);  $\leq$ 1% d<sub>0</sub> Incorporation:

UV/Vis.:  $\lambda_{\text{max}}$ : 225 nm Supplied as: A crystalline solid

-20°C Storage:

Stability: As supplied, 2 years from the QC date provided on the Certificate of Analysis, when

stored properly

### **Laboratory Procedures**

Aspirin-d<sub>4</sub> contains 4 deuterium atoms at the 2, 3, 4, and 5 positions. It is intended for use as an internal standard for the quantification of aspirin (Item No. 70260) by GC- or LC-MS. The accuracy of the sample weight in this vial is between 5% over and 2% under the amount shown on the vial. If better precision is required, the deuterated standard should be quantitated against a more precisely weighed unlabeled standard by constructing a standard curve of peak intensity ratios (deuterated versus unlabeled).

Aspirin- $d_A$  is supplied as a crystalline solid. A stock solution may be made by dissolving the aspirin- $d_A$  in the solvent of choice. Aspirin- $d_4$  is soluble in organic solvents such as ethanol, DMSO, and dimethyl formamide, which should be purged with an inert gas. The solubility of aspirin-d₁ in these solvents is approximately 80, 41, and 30 mg/ml, respectively.

Further dilutions of the stock solution into aqueous buffers or isotonic saline should be made prior to performing biological experiments. Ensure that the residual amount of organic solvent is insignificant, since organic solvents may have physiological effects at low concentrations. Organic solvent-free aqueous solutions of aspirin-d<sub>4</sub> can be prepared by directly dissolving the crystalline solid in aqueous buffers. The solubility of aspirin- $d_{\Delta}$  in PBS, pH 7.2, is approximately 2.7 mg/ml. Avoid adding aspirin to basic solutions (pH > 7.4), since base treatment will hydrolyze aspirin to salicylic acid. Store aqueous solutions of aspirin on ice and use within 30 minutes of preparation.

#### Description

Aspirin is a non-selective, irreversible COX inhibitor. The  $\rm IC_{50}$  values for ovine COX-1 and -2 are 0.75 and 1.25 mM, respectively. Aspirin acetylates COX-1 at Ser $^{530}$  and COX-2 at Ser $^{516}$  resulting in irreversible enzyme inhibition.

### Reference

1. Johnson, J.L., Wimsatt, J., Buckel, S.D., et al. Purification and characterization of prostaglandin H synthase-2 from sheep placental cotyledons. Arch. Biochem. Biophys. 324, 26-34 (1995).

WARNING
THIS PRODUCT IS FOR RESEARCH ONLY - NOT FOR HUMAN OR VETERINARY DIAGNOSTIC OR THERAPEUTIC USE.

This material should be considered hazardous until further information becomes available. Do not ingest, inhale, get in eyes, on skin, or on clothing. Wash thoroughly after handling. Before use, the user must review the complete Safety Data Sheet, which has been sent via email to your institution.

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