

Produktinformation



Forschungsprodukte & Biochemikalien



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PRODUCT INFORMATION



Aniline-d₅ Item No. 35131

CAS Registry No.: 4165-61-1

Formal Name: benzen-2,3,4,5,6-d₅-amine

Synonyms: Aminobenzene-d₅, Phenyl-d₅ amine,

Benzenamine-d₅

MF: $C_6H_2D_5N$ 98.2 FW:

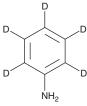
Chemical Purity: ≥98% (Aniline)

Deuterium

Incorporation: \geq 99% deuterated forms (d₁-d₅); \leq 1% d₀

UV/Vis.: λ_{max} : 235 nm Supplied as: A liquid -20°C Storage: Stability: ≥1 year

Information represents the product specifications. Batch specific analytical results are provided on each certificate of analysis.



Laboratory Procedures

Aniline- d_5 is intended for use as an internal standard for the quantification of aniline by GC- or LC-MS. The accuracy of the sample weight in this vial is between 5% over and 2% under the amount shown on the vial. If better precision is required, the deuterated standard should be quantitated against a more precisely weighed unlabeled standard by constructing a standard curve of peak intensity ratios (deuterated versus unlabeled).

Aniline- d_5 is supplied as a liquid. A stock solution may be made by dissolving the aniline- d_5 in the solvent of choice, which should be purged with an inert gas. Aniline- d_5 is soluble in organic solvents such as ethanol, DMSO, and dimethyl formamide. The solubility of aniline-d₅ in these solvents is approximately 3, 10, and 15 mg/ml, respectively.

Description

Aniline is an aromatic amine that has been used as a starting material in the synthesis of a wide variety of precursors to dyes and pigments, rubber processing chemicals, and agricultural chemicals. It increases the levels of reactive oxygen species (ROS) and malondialdehyde (MDA) and decreases the levels of glutathione and catalase, as well as the activity of superoxide dismutase (SOD), in primary rat hepatocytes when used at concentrations of 5 and 10 μg/ml.² Aniline (0.3 and 0.6% of the diet) induces splenic sarcomas in rats and is neurotoxic to rats when administered at doses of 750 and 1,000 mg/kg.^{3,4} Formulations containing aniline have been used in the synthesis of pharmaceuticals.

References

- 1. Kahl, T., Schröder, K.-Q., Lawrence, F.R., et al. Aniline. Ullmann's Encyclopedia of Industrial Chemistry. 6th edition (2011).
- 2. Wang, Y., Gao, H., Na, X.-L., et al. Aniline induces oxidative stress and apoptosis of primary cultured hepatocytes. Int. J. Environ. Res. Public Health 13(12), 1188 (2016).
- Goodman, D.G., Ward, J.M., and Reichardt, W.D. Splenic fibrosis and sarcomas in F344 rats fed diets containing aniline hydrochloride, p-chloroaniline, azobenzene, o-toluidine hydrochloride, 4,4'-sulfonyldianiline, or D & C red No. 9. J. Natl. Cancer Inst. 73(1), 265-273 (2019).
- 4. Okazaki, Y., Yamashita, K., Sudo, M., et al. Neurotoxicity induced by a single oral dose of aniline in rats. J. Vet. Med. Sci. 63(5), 539-546 (2001).

WARNING
THIS PRODUCT IS FOR RESEARCH ONLY - NOT FOR HUMAN OR VETERINARY DIAGNOSTIC OR THERAPEUTIC USE.

This material should be considered hazardous until further information becomes available. Do not ingest, inhale, get in eyes, on skin, or on clothing. Wash thoroughly after handling. Before use, the user must review the complete Safety Data Sheet, which has been sent via email to your institution.

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