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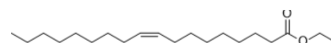
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Ethyl oleate

Cat. No.:	HY-N7103
CAS No.:	111-62-6
Molecular Formula:	C ₂₀ H ₃₈ O ₂
Molecular Weight:	310.51
Target:	Endogenous Metabolite
Pathway:	Metabolic Enzyme/Protease
Storage:	<div>Pure form -20°C 3 years</div> <div>In solvent -80°C 6 months</div> <div> -20°C 1 month</div>



SOLVENT & SOLUBILITY

In Vitro	DMSO : 50 mg/mL (161.03 mM; Need ultrasonic)					
	Preparing Stock Solutions	<div><div>Solvent</div><div>Concentration</div></div>	Mass	1 mg	5 mg	10 mg
		1 mM		3.2205 mL	16.1025 mL	32.2051 mL
		5 mM		0.6441 mL	3.2205 mL	6.4410 mL
		10 mM		0.3221 mL	1.6103 mL	3.2205 mL
Please refer to the solubility information to select the appropriate solvent.						
In Vivo	1. Add each solvent one by one: 10% DMSO >> 40% PEG300 >> 5% Tween-80 >> 45% saline Solubility: ≥ 2.5 mg/mL (8.05 mM); Clear solution					
	2. Add each solvent one by one: 10% DMSO >> 90% (20% SBE-β-CD in saline) Solubility: 2.5 mg/mL (8.05 mM); Suspended solution; Need ultrasonic					
	3. Add each solvent one by one: 10% DMSO >> 90% corn oil Solubility: ≥ 2.5 mg/mL (8.05 mM); Clear solution					

BIOLOGICAL ACTIVITY

Description	Ethyl oleate is an orally active fatty acid ester formed from the condensation of oleic acid and ethanol. Ethyl oleate is the main fatty acid ethyl ester in the blood after alcohol ingestion. Ethyl oleate has no obvious toxicity to rats and its absorption, distribution and excretion are similar to triacylglycerol. Ethyl oleate can accelerate the drying process of certain foods and can also be used as a liquid lipid component in nanostructured lipid carriers ^{[1][2][3][4]} .
In Vitro	<p>Fatty acid ethyl esters (FAEEs) can be synthesized by adding ethanol in Hep G2 cells, in which Ethyl palmitate (HY-N2086) and Ethyl oleate are the main FAEEs^[1].</p> <p>MCE has not independently confirmed the accuracy of these methods. They are for reference only.</p>

In Vivo

Ethyl oleate (oral gavage; 1.7-3.4g /kg; single dose) has no obvious toxicity to rats and good absorption, mainly distributes in mesenteric fat and excretes as CO₂, is similar to triacylglycerol (Ethyl oleate radiolabelled)^[2].
MCE has not independently confirmed the accuracy of these methods. They are for reference only.

REFERENCES

- [1]. Dan L, et al. Ethyl palmitate and ethyl oleate are the predominant fatty acid ethyl esters in the blood after ethanol ingestion and their synthesis is differentially influenced by the extracellular concentrations of their corresponding fatty acids. *Alcohol Clin Exp Res.* 1997 Apr;21(2):286-92.
- [2]. Bookstaff RC, et al. The safety of the use of ethyl oleate in food is supported by metabolism data in rats and clinical safety data in humans. *Regul Toxicol Pharmacol.* 2003 Feb;37(1):133-48.
- [3]. Saravacos G D, et al. Effect of ethyl oleate on the rate of air-drying of foods. *Journal of Food Engineering*, 1988, 7(4): 263-270.
- [4]. Zhang Y, et al. Ethyl oleate-containing nanostructured lipid carriers improve oral bioavailability of trans-ferulic acid as compared with conventional solid lipid nanoparticles. *Int J Pharm.* 2016 Sep 10;511(1):57-64.
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Caution: Product has not been fully validated for medical applications. For research use only.

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