Isopentyl nitrite



Section 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

PRODUCT NAME

Isopentyl nitrite

STATEMENT OF HAZARDOUS NATURE

CONSIDERED A HAZARDOUS SUBSTANCE ACCORDING TO OSHA 29 CFR 1910.1200.



SUPPLIER

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SYNONYMS

C5-H11-NO2, (CH3)2CHCH2CH2ONO, amilnitrit, "amyl nitrit", "amyl nitrit", "iso-pentyl nitrite", "3-methylbutanol nitrite", "3-methylbutyl nitrite", "iso-pentyl nitrite", "iso-pentyl nitrite", "3-methylbutyl ester", Aspiral, IPN, Nitramyl, Vaporole





EMERGENCY OVERVIEW RISK

Harmful by inhalation and if swallowed. Irritating to eyes and skin. Highly flammable.

POTENTIAL HEALTH EFFECTS

ACUTE HEALTH EFFECTS

SWALLOWED

• Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.

EYE

This material can cause eye irritation and damage in some persons.

SKIN

- This material can cause inflammation of the skin oncontact in some persons.
- The material may accentuate any pre-existing dermatitis condition.
- Skin contact with the material may damage the health of the individual; systemic effects may result following absorption.
- Open cuts, abraded or irritated skin should not be exposed to this material.

• Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

INHALED

■ Inhalation of vapors or aerosols (mists, fumes), generated by the material during the course of normal handling, may be harmful.

■ The material is not thought to produce respiratory irritation (as classified using animal models). Nevertheless inhalation of vapors, fumes or aerosols, especially for prolonged periods, may produce respiratory discomfort and occasionally, distress.

■ Inhalation of volatile nitrites produces vasodilation and hypotension lasting about 90 seconds, occasionally accompanied by swooning (posture-related syncope). Pulsating headache, rapid flushing of the face and dizziness are common. Confusion, vertigo, motor restlessness, weakness, cyanosis, nausea and vomiting may occur in some individuals.

Absorption into the body leads to the formation of methaemoglobin which in sufficient concentration causes cyanosis. Onset may be delayed for several hours.

■ The substance and/or its metabolites may bind to hemoglobin inhibiting normal uptake of oxygen. This condition, known as "methemoglobinemia", is a form of oxygen starvation (anoxia).

CHRONIC HEALTH EFFECTS

■ Long-term exposure to the product is not thought to produce chronic effects adverse to the health (as classified using animal models); nevertheless exposure by all routes should be minimized as a matter of course.

Chronic abuse of amyl nitrite has resulted in Heinz body haemolytic anaemia in two individuals.

S	Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS			
NAME	CAS RN	%		
iso-amyl nitrite	110-46-3	>98		

Section 4 - FIRST AID MEASURES

SWALLOWED

· IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY. · Where Medical attention is not immediately available or where the patient is more than 15 minutes from a hospital or unless instructed otherwise:

EYE

■ If this product comes in contact with the eyes: • Wash out immediately with fresh running water. • Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.

SKIN

■ If skin contact occurs: · Immediately remove all contaminated clothing, including footwear · Flush skin and hair with running water (and soap if available).

INHALED

· If fumes or combustion products are inhaled remove from contaminated area. · Lay patient down. Keep warm and rested.

NOTES TO PHYSICIAN

For lower alkyl nitrites:

Consider treatment regime for amyl nitrate:

Synacope should be treated by keeping the patient in a recumbent position with the head lowered. Severe poisoning should be treated by

gastric aspiration and lavage and the administration of oxygen with assisted respiration if necessary. If methaemoglinaemia occurs give methylene blue 1 to 4 mg body-weight by intravenous injection. The circulation may be maintained with infusions of plasma or suitable electrolyte solutions

Martindale.

The material may induce methemoglobinemia following exposure.

Initial attention should be directed at oxygen delivery and assisted ventilation if necessary. Hyperbaric oxygen has not demonstrated substantial benefits.

· Hypotension should respond to Trendelenburg's position and intravenous fluids; otherwise dopamine may be needed.

Section 5 - FIRE FIGHTING MEASURES

Vapour Pressure (mmHG):	Not available
Upper Explosive Limit (%):	Not available
Specific Gravity (water=1):	0.875
Lower Explosive Limit (%):	Not available

EXTINGUISHING MEDIA

· Foam.

· Dry chemical powder.

FIRE FIGHTING

· Alert Emergency Responders and tell them location and nature of hazard.

May be violently or explosively reactive.

When any large container (including road and rail tankers) is involved in a fire,

consider evacuation by 1000 metres in all directions.

GENERAL FIRE HAZARDS/HAZARDOUS COMBUSTIBLE PRODUCTS

· Liquid and vapor are highly flammable.

· Severe fire hazard when exposed to heat, flame and/or oxidizers.

Combustion products include: carbon dioxide (CO2), nitrogen oxides (NOx), other pyrolysis products typical of burning organic material. **FIRE INCOMPATIBILITY**

FIRE INCOMPATIBILITY

Avoid contamination with oxidizing agents i.e. nitrates, oxidizing acids, chlorine bleaches, pool chlorine etc. as ignition may result.

PERSONAL PROTECTION

Glasses: Chemical goggles. Gloves: 1.PVA 2.NITRILE 3.VITON Respirator: Type A-P Filter of sufficient capacity

Section 6 - ACCIDENTAL RELEASE MEASURES

MINOR SPILLS

- · Remove all ignition sources.
- · Clean up all spills immediately.
- MAJOR SPILLS
- · Clear area of personnel and move upwind.
- · Alert Emergency Responders and tell them location and nature of hazard.

Section 7 - HANDLING AND STORAGE

PROCEDURE FOR HANDLING

- \cdot Containers, even those that have been emptied, may contain explosive vapours.
- · Do NOT cut, drill, grind, weld or perform similar operations on or near containers.
- · DO NOT allow clothing wet with material to stay in contact with skin.
- · Avoid all personal contact, including inhalation.
- · Wear protective clothing when risk of exposure occurs.

RECOMMENDED STORAGE METHODS

Glass container.

Packing as supplied by manufacturer. Plastic containers may only be used if approved for flammable liquid.

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· For low viscosity materials (i): Drums and jerricans must be of the non-removable head type. (ii): Where a can is to be used as an inner package, the can must have a screwed enclosure.

· For materials with a viscosity of at least 2680 cSt. (23 deg. C).

STORAGE REQUIREMENTS

· Store in original containers in approved flame-proof area.

· No smoking, naked lights, heat or ignition sources.

Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

EXPOSURE CONTROLS

The following materials had no OELs on our records • iso-amyl nitrite: CAS:110-46-3

PERSONAL PROTECTION



RESPIRATOR

Type A-P Filter of sufficient capacity Consult your EHS staff for recommendations

EYE

· Safety glasses with side shields.

· Chemical goggles.

HANDS/FEET

■ Wear chemical protective gloves, eg. PVC.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include: such as:

· frequency and duration of contact,

· chemical resistance of glove material,

· glove thickness and

· dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739).

• When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374) is recommended.

 \cdot When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374) is recommended.

· Contaminated gloves should be replaced.

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

OTHER

- · Overalls.
- · PVC Apron.

· Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity.

· For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets), non sparking safety footwear.

ENGINEERING CONTROLS

■ For flammable liquids and flammable gases, local exhaust ventilation or a process enclosure ventilation system may be required. Ventilation equipment should be explosion-resistant.

Section 9 - PHYSICAL AND CHEMICAL PROPERTIES

PHYSICAL PROPERTIES

Liquid. Does not mix with water. Floats on water.			
State	Liquid	Molecular Weight	117.15
Melting Range (°F)	Not available	Viscosity	Not Available
Boiling Range (°F)	206.6- 210.2	Solubility in water (g/L)	Partly miscible
Flash Point (°F)	50	pH (1% solution)	Not applicable
Decomposition Temp (°F)	Not Available	pH (as supplied)	Not applicable
Autoignition Temp (°F)	404.6	Vapour Pressure (mmHG)	Not available
Upper Explosive Limit (%)	Not available	Specific Gravity (water=1)	0.875
Lower Explosive Limit (%)	Not available	Relative Vapor Density (air=1)	Not available
Volatile Component (%vol)	100	Evaporation Rate	Not available

APPEARANCE

Yellow, transparent liquid with fragrant, penetrating, fruity odour and pungent aromatic taste; does not mix well with water. Soluble in alcohol, chloroform and ether. Decomposes on exposure to air, light or water.

Section 10 - CHEMICAL STABILITY

CONDITIONS CONTRIBUTING TO INSTABILITY

 \cdot Presence of incompatible materials.

· Product is considered stable.

STORAGE INCOMPATIBILITY

For alkyl nitrites:

 \cdot Many alkyl nitrites are thermally unstable and may readily decompose or explode on heating. Lower alkyl nitrites have been known to decompose and burst their container even under refrigerated storage conditions.

BRETHERICK L.: Handbook of Reactive Chemical Hazards

· Prolonged exposure to light, air or water may cause decomposition producing oxides of nitrogen.

• The alkyl nitrites are strong oxidisers and react violently with reducing agents, combustibles, organics and other readily oxidisable materials. (particularly finely divided metals)

· Segregate from alcohols, antipyrine, alkaline materials, alkaline carbonates, potassium iodide, bromides,, and ferrous salts.

· May corrode metal in the presence of moisture.

Avoid reaction with oxidizing agents.

For incompatible materials - refer to Section 7 - Handling and Storage.

Section 11 - TOXICOLOGICAL INFORMATION

ISO-AMYL NITRITE

TOXICITY AND IRRITATION ISO-AMYL NITRITE:

■ unless otherwise specified data extracted from RTECS - Register of Toxic Effects of Chemical Substances.

TOXICITY	IRRITATION
Oral (rat) LD50: 505 mg/kg	Nil Reported
Inhalation (rat) LD50: 716 ppm/4h	
Inhalation (mouse) LD50: 130 mg/kg	

Section 12 - ECOLOGICAL INFORMATION

This material and its container must be disposed of as hazardous waste.

Ecotoxicity

Ingredient	Persistence: Water/Soil	Persistence: Air	Bioaccumulation	Mobility
iso-amyl nitrite	HIGH		LOW	MED

Section 13 - DISPOSAL CONSIDERATIONS

US EPA Waste Number & Descriptions

A. General Product Information

Ignitability characteristic: use EPA hazardous waste number D001 (waste code I)

Disposal Instructions

All waste must be handled in accordance with local, state and federal regulations.

Puncture containers to prevent re-use and bury at an authorized landfill.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- Reduction
- · Reuse
- · Recycling
- · Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

DO NOT allow wash water from cleaning equipment to enter drains. Collect all wash water for treatment before disposal. • Recycle wherever possible.

· Consult manufacturer for recycling options or consult Waste Management Authority for disposal if no suitable treatment or disposal facility can be identified.

Section 14 - TRANSPORTATION INFORMATION

DOT:

Symbols: None Hazard class or Division: 3 Identification Numbers: UN1113 PG: II Label Codes: 3 Special provisions: IB2, T4, TP1 Packaging: Exceptions: 150 Packaging: Non- bulk: 202 Packaging: Exceptions: 150 Quantity limitations: 5 L Passenger aircraft/rail: Quantity Limitations: Cargo 60 L Vessel stowage: Location: E aircraft only: Vessel stowage: Other: 40 Hazardous materials descriptions and proper shipping names: Amyl nitrites Air Transport IATA: ICAO/IATA Class: 3 ICAO/IATA Subrisk: None UN/ID Number: 1113 Packing Group: II Special provisions: None Cargo Only Packing Instructions: 307 Maximum Qty/Pack: 60 L Passenger and Cargo Passenger and Cargo Packing Instructions: 305 Maximum Qty/Pack: 5 L Passenger and Cargo Limited Quantity Passenger and Cargo Limited Quantity Packing Instructions: Y305 Maximum Qty/Pack: 1 L Shipping Name: AMYL NITRITE

Maritime Transport IMDG:

IMDG Class: 3 IMDG Subrisk: None UN Number: 1113 Packing Group: II EMS Number: F-E , S-D Special provisions: None Limited Quantities: 1 L Shipping Name: AMYL NITRITE

Section 15 - REGULATORY INFORMATION

iso-amyl nitrite (CAS: 110-46-3) is found on the following regulatory lists;

"Canada Domestic Substances List (DSL)","Canada Toxicological Index Service - Workplace Hazardous Materials Information System - WHMIS (English)","US - Massachusetts Oil & Hazardous Material List","US - New Jersey Right to Know Hazardous Substances","US DOE Temporary Emergency Exposure Limits (TEELs)","US Toxic Substances Control Act (TSCA) - Inventory"

Section 16 - OTHER INFORMATION

LIMITED EVIDENCE

■ Skin contact may produce health damage*.

* (limited evidence).

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Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references. A list of reference resources used to assist the committee may be found at: www.chemwatch.net/references

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• The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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